

# Semiempirical Method MNDO for the Evaluation of the Effect of Different Substituents at the Imine-Carbon Position on the Acetaldemine-Vinylamine Tautomerization and Comparison to the Substitution at $\alpha$ -Position

Hamzeh S. M. Al-Omari

Chemistry Department, College of Science, University of Mutah, Karak, Jordan

Reprint requests to Dr. H. S. M. Al-O. E-mail: hamzehs@mutah.edu.jo

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MNDO calculations have been employed to study the effect of some substituents of the acetaldemine-vinylamine tautomeric system at the imine-carbon position of  $\text{CH}_3\text{CXNH}$ , where  $\text{X} = \text{H}, \text{F}, \text{CN}, \text{NH}_2, \text{NO}_2, \text{BH}_2, \text{CH}_3$  and  $\text{CF}_3$ . It is found that the substituents  $\text{F}, \text{NH}_2$  and  $\text{NO}_2$  encourage the formation of the enamine tautomer. The substituents  $\text{CN}, \text{CH}_3, \text{CF}_3$  and  $\text{BH}_2$  encourage the formation of the imine tautomer. Isodesmic reactions, free energy change ( $\Delta G$ ), charge distribution and energy gap ( $E_g$ ) between HOMO and LUMO were used to prove these findings. Resonance stabilization was a major factor in the determination of the most stable tautomer. These results were compared with previous work on the substitution at the  $\sigma$ -carbon position.

*Key words:* Acetaldemine; Vinylamine; Tautomerism; Imine/enamine; Resonance; MNDO.